



Total Questions : 50

Time : 1 hr.

PATTERN & MARKING SCHEME			
Section	(1) Physics & Chemistry	(2) Achievers Section	(3) Mathematics or Biology
No. of Questions	25	5	20
Marks per Ques.	1	3	1



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**SYLLABUS**

**Section – 1 : Physics :** Electricity and Magnetism, Electromagnetic Induction, Alternating current, Electromagnetic waves, Optics, Modern Physics, Semiconductor Electronics.  
**Chemistry :** Solutions, Electrochemistry, Chemical Kinetics, The *d*- and *f*-Block Elements, Coordination Compounds, Haloalkanes and Haloarenes, Alcohols, Phenols and Ethers, Aldehydes, Ketones and Carboxylic Acids, Amines, Biomolecules.

**Section – 2 :** Higher Order Thinking Questions - Syllabus as per Section – 1.

**Section – 3 :** Relations and Functions, Inverse Trigonometric Functions, Matrices and Determinants, Continuity and Differentiability, Application of Derivatives, Integrals, Application of Integrals, Differential Equations, Vector Algebra, Three Dimensional Geometry, Probability, Linear Programming.

OR

**Section – 3 :** Reproduction, Genetics and Evolution, Biology in Human Welfare, Biotechnology, Ecology.

**PHYSICS AND CHEMISTRY**

1. A ray of light passes from vacuum into a medium of refractive index  $\mu$ . If the angle of incidence is found to be twice the angle of refraction, then the angle of incidence is

- (A)  $\cos^{-1}(\mu/2)$                       (B)  $2\cos^{-1}(\mu/2)$   
(C)  $2\sin^{-1}\mu$                           (D)  $2\sin^{-1}(\mu/2)$ .

2. A capacitor of capacitance  $C_0$  is charged to a potential  $V_0$  and then isolated. A small capacitor of capacitance  $C$  is then charged from  $C_0$ , discharged and charged again, the process being repeated  $n$  times. Due to this, potential of the larger capacitor is decreased to  $V$ . The value of  $C$  is

- (A)  $C_0 \left[ \frac{V_0}{V} \right]^{1/n}$                       (B)  $C_0 \left[ \left( \frac{V_0}{V} \right)^{1/n} - 1 \right]$   
(C)  $C_0 \left[ \left( \frac{V_0}{V} \right) - 1 \right]^n$                       (D)  $C_0 \left[ \left( \frac{V_0}{V} \right)^n - 1 \right]$ .

3. A ray of light in a liquid of refractive index 1.4, approaches the boundary surface between the liquid and air at an angle of incidence whose sine is 0.8. Which of the following statements is correct about the behavior of the light ?

- (A) It is impossible to predict the behavior of the light ray on the basis of the information supplied.  
(B) The sine of the angle of refraction of the emergent ray will be less than 0.8.

- (C) The ray will be internally reflected.  
(D) None of these

4. Five aromatic compounds are given as :

- I. *p*-Methoxybenzaldehyde  
II. *p*-Hydroxyphenylacetaldehyde  
III. *p*-Hydroxyphenylmethyl ketone  
IV. *p*-Methylbenzoic acid  
V. *p*-Hydroxyphenylvinyl ether

Now, read the given passage and fill in the blanks by selecting an appropriate option.

Compound(s) (p) form(s) a silver mirror with Tollens' reagent. Compound (q) gives positive iodoform test while compound (r) on acid hydrolysis gives 1, 4-dihydroxybenzene. Compound (s) can be readily extracted in aqueous  $\text{NaHCO}_3$  solution.

- |     | (p)    | (q) | (r) | (s) |
|-----|--------|-----|-----|-----|
| (A) | I, II  | V   | IV  | II  |
| (B) | I      | IV  | III | V   |
| (C) | III, V | IV  | V   | II  |
| (D) | I, II  | III | V   | IV  |

5. 1.00 g of a non-electrolyte solute dissolved in 50 g of benzene lowered the freezing point of benzene by 0.40 K. The freezing point depression constant of benzene is  $5.12 \text{ K kg mol}^{-1}$ . The molar mass of the solute is

- (A) 456 kg/mol                              (B) 352 g/mol  
(C) 256 g/mol                              (D) 240 g/mol.

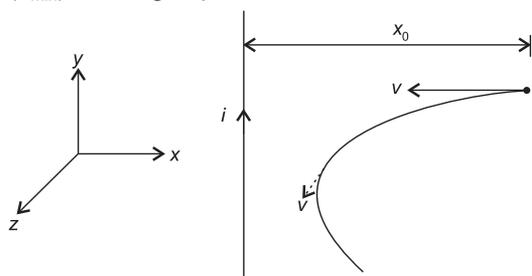
6. When a compound 'X' is reacted with  $\text{PCl}_5$  and then with  $\text{NH}_3$ , it gives 'Y'. When 'Y' is treated with  $\text{Br}_2$  and  $\text{KOH}$ , it produces 'Z'. 'Z' on treatment with  $\text{NaNO}_2$  and  $\text{HCl}$  at  $0^\circ\text{C}$  and then on warming with

water produces *ortho*-cresol. Compound 'X' is

- (A) *o*-toluic acid (B) *o*-chlorotoluene  
(C) *o*-bromotoluene (D) *m*-toluic acid.

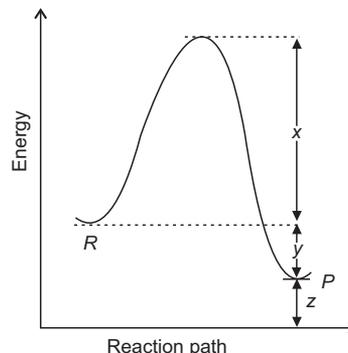
### ACHIEVERS SECTION

7. A long straight wire carries a current  $i$ . A particle having a positive charge  $q$  and mass  $m$ , kept at distance  $x_0$  from the wire is projected towards it with speed  $v$  as shown in the figure. The closest distance of approach ( $x_{\min}$ ) of charged particle to the wire is



- (A)  $x_{\min} = x_0 e^{-2\pi m v / \mu_0 q i}$   
(B)  $x_{\min} = x_0 e^{m v / \mu_0 q i}$   
(C)  $x_{\min} = x_0 e^{-\pi m v / \mu_0 q i}$   
(D)  $x_{\min} = x_0 e^{-m v / \mu_0 q 2 i}$

8. Observe the given graph carefully.



The activation energy of the backward reaction, heat of reaction and threshold energy of the reaction respectively are

- (A)  $x - y$ ,  $y$  and  $x + y - z$   
(B)  $x + y + z$ ,  $y + z$  and  $z$   
(C)  $x + y$ ,  $y$  and  $x + y + z$   
(D)  $x + y$ ,  $y$  and  $x - y - z$

### MATHEMATICS

9. If  $\sqrt{1-x^6} + \sqrt{1-y^6} = a(x^3 - y^3)$  and

$$\frac{dy}{dx} = f(x, y) \sqrt{\frac{1-y^6}{1-x^6}}, \text{ then}$$

- (A)  $f(x, y) = \frac{y}{x}$  (B)  $f(x, y) = \frac{y^2}{x^2}$   
(C)  $f(x, y) = \frac{2y^2}{x^2}$  (D)  $f(x, y) = \frac{x^2}{y^2}$

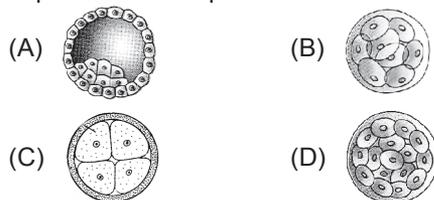
10. For non-zero vectors  $\vec{a}$ ,  $\vec{b}$ ,  $\vec{c}$ ;

$$(\vec{a} \times \vec{b}) \cdot \vec{c} = |\vec{a}| |\vec{b}| |\vec{c}| \text{ holds iff}$$

- (A)  $\vec{a} \cdot \vec{b} = 0, \vec{b} \cdot \vec{c} = 0$   
(B)  $\vec{b} \cdot \vec{c} = 0, \vec{c} \cdot \vec{a} = 0$   
(C)  $\vec{c} \cdot \vec{a} = 0, \vec{a} \cdot \vec{b} = 0$   
(D)  $\vec{a} \cdot \vec{b} = \vec{b} \cdot \vec{c} = \vec{c} \cdot \vec{a} = 0$

### BIOLOGY

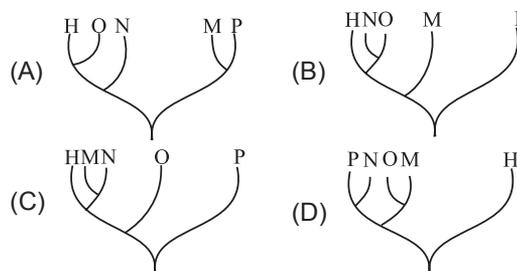
9. In which of the following embryonic stages does the implantation take place?



10. Human blood, when mixed with antibodies present in human blood, give maximum precipitation. If another animal's blood is mixed with antibodies present in human blood, the percentage of precipitation indicates evolutionary relationship with that animal.

The following experimental results were obtained :  
Species: Human (H) - 100%; M - 37%; N - 75%;  
O - 79%; P - 17%

Which phylogenetic tree would best represent these results?



### ANSWERS

- NSO - (PHYSICS AND CHEMISTRY) 1. (B) 2. (B) 3. (C) 4. (D) 5. (C) 6. (A) 7. (A) 8. (C)  
(MATHEMATICS) 9. (D) 10. (D)  
(BIOLOGY) 9. (A) 10. (A)